



# ID Card

## Palladium dichloride

Version 19 December 2016

**Notes:**

- This ID card is used to support the substance sameness discussions in SIEFs and to describe the substance to the best of the SIEF members' knowledge.
- It also aims at grouping communications relevant to the request of available data or information, the approval of the proposed Lead Registrant and the registration strategy with the SIEF.
- It is the responsibility of each individual registrant to identify their substance and to report company-specific identity in their Registration Dossier (section 1 of IUCLID).

**DISCLAIMER**

All data and information contained in this document shall be treated by the receiving party (i) in full confidence with the adequate respect of any confidential and/or proprietary nature of such information and (ii) only in the framework of the purpose of agreeing on substance sameness, Lead Registrant and overall REACH Strategy for the concerned Substance under REACH (the 'Purpose').

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### 1. Identification of the substance

**Table 1. Identification of the substance**

	<b>Original (in EC inventory)</b>
<b>Name</b>	Palladium dichloride
<b>EC number</b>	231-596-2
<b>CAS number</b>	7647-10-1
<b>Description</b>	Not available
<b>Composition type</b>	Mono-constituent substance

### 2. Synonyms and other identifiers of the substance

**Table 2. Synonyms and other identifiers of the substance**

<b>IUPAC name</b>	Palladium(2+) dichloride
<b>CAS name</b>	Palladium chloride
<b>Abbreviations</b>	
<b>Other commercial or international names</b>	Palladium chloride Palladium (II) dichloride Dichloropalladium Palladium(2+) chloride Palladium chloride (2+)
<b>Other identity codes</b>	None



### 3. Substances (with core identifiers) also falling under this substance (with justification)

None

### 4. Information related to molecular and structural formula of the substance

Table 3. Information related to molecular and structural formula of the substance

<b>Molecular formula</b>	Cl <sub>2</sub> Pd
<b>Structural formula</b>	
<b>Smiles notation</b>	[Cl-].[Cl-].[Pd+2]
<b>Optical activity</b>	Not applicable
<b>Typical ratio of (stereo) isomers</b>	Not available
<b>Molecular Weight / Molecular Weight range</b>	177,33 g/mol

### 5. Typical composition of the substance

Table 4. Typical composition

	Name	Symbol / Formula	Min & Max concentrations (%) <sup>§</sup>	Typical concentration (%) <sup>§§</sup>
<b>Main constituent(s)*</b>	Palladium dichloride	Cl <sub>2</sub> Pd	99 - 100 <sup>§</sup>	> 99
<b>Impurity(ies)<sup>#</sup></b>	Several minor (especially metallic) impurities which do not affect the classification of the substance because of their non-hazardous nature or because they do not exceed the classification cut-off limits in the substance	e.g. Ag, Au, Cu, Ir, Pb, Pt, Rh, Ru	0 - 1	< 1

\* ≥ 80 % (w/w) for mono-constituent substances; ≥ 10 % (w/w) and < 80 % (w/w) for multi-constituent substances.

<sup>#</sup> An impurity is an unintended constituent present in a substance, as produced. It may originate from the starting materials or be the result of secondary or incomplete reactions during the production process. While impurities are present in the final substance, they were not intentionally added.

<sup>§</sup> Concentration ranges define the substance sameness criteria agreed by all Consortium Members in preparation of the communication with other SIEF members.

<sup>§§</sup> Typical concentration refers to the representative sample used for testing.

<sup>§</sup> Corresponds to 59,4 - 60 % Pd.

The composition given above is typical and should therefore represent the majority of Palladium dichloride as manufactured and/or imported in the EEA market. Palladium dichloride containing less than 99 % Palladium dichloride may still be considered to be the same for the purpose of registration under REACH and may be referred to as impure Palladium dichloride to distinguish it from the typically pure Palladium dichloride.



## 6. Information on appearance, physical state and properties of the substance

**Table 5. Appearance / physical state / properties of the solid substance**

<b>Physical state</b>	Solid
<b>Physical form*</b>	Crystalline
<b>Appearance</b>	Red to brown powder
<b>Particle size**</b>	Fine to coarse powder
<b>Does the solid hydrolyse?#</b>	No
<b>Is the solid hygroscopic?§</b>	Yes

\* Crystalline form: solid material whose constituent atoms, molecules, or ions are arranged in an ordered pattern extending in all three spatial dimensions. Amorphous form: solid material whose constituent atoms, molecules, or ions are randomly arranged.

\*\* Nanoform: particles in the size range 1 - 100 nm (for full definition of a nanomaterial, see <http://ec.europa.eu/environment/chemicals/nanotech/index.htm#definition>). Fine powder: particles in the size range 100 – 2.500 nm. Coarse powder: particles in the size range 2.500 nm – 1 mm. Massive object: particles in the size range > 1 mm.

# Hydrolysis: decomposition (cleavage of chemical bonds) by the addition of water.

§ Hygroscopic substance: readily attracts water from its surroundings, through either absorption or adsorption. Cf. also water/moisture content in Table 4.

## 7. Analytical data

Annex VI of REACH requires the registrant to describe the analytical methods and/or to provide the bibliographical references for the methods used for identification of the substance and, where appropriate, for the identification of impurities and additives. This information should be sufficient to allow the methods to be reproduced.

**Table 6. Analytical methods for identification of the substance**

Parameter / Method	Recommended for substance identification and sameness check	Applicable	Not applicable or not recommended
<b>Elemental analysis</b>			
ICP (ICP-MS or ICP-OES)	X		
Atomic absorption spectroscopy (AAS)			
Glow discharge mass spectrometry (GDMS)			
<b>Molecular analysis</b>			
Infrared (IR) spectroscopy	X		
Raman spectroscopy			
<b>Mineralogical analysis</b>			
X-Ray Fluorescence (XRF)		X	
X-Ray Diffraction (XRD)	X		
<b>Morphology and particle sizing</b>			
Electron microscopy (SEM, TEM, REM)* #			
Laser diffraction* #	X		
Particle size by other means (e.g. sieve analysis)#			
Surface area by N-BET* #	X		
<b>Other</b>			



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\* Analytical techniques particularly (but not exclusively) relevant for nanomaterials.

# The choice of the technique for particle size depends on the size of the material as manufactured/imported/placed on the market/used.

### 8. Lead Registrant

BASF (Italia) volunteers to be the Lead Registrant for Palladium dichloride. The EPMF will provide support to the Lead Registrant as laid down in the EPMF Agreement.

### 9. REACH Strategy

The table below presents the overall Registration Strategy for Palladium dichloride based on the information available to the EPMF by the date given above on the document.

The Registration Dossier will be prepared for the highest substance status (information requirements associated to a substance or Article 10 Registration being higher than an intermediate handled under strictly controlled conditions or Article 17 or 18 one) and associated tonnage band.

The recap below therefore reflects the scope of work of the EPMF for Palladium dichloride and sets the minimum and maximum set of information that will be gathered and/or produced when preparing the Registration Dossier for Palladium dichloride as described in this ID Card.

If higher information requirements are necessary, these can be included in the Registration dossier (if EPMF is made aware of these additional requirements in-time) as an update to the already submitted dossier.

**Table 7. REACH strategy for the substance (basis for REACH Registration preparation)**

Item	Description
<b>REACH category</b>	Mono-constituent substance
<b>Highest status</b>	Substance
<b>Highest tonnage band</b>	10-100 t/a
<b>Information requirements</b>	Available / Existing + Annex VII + Annex VIII
<b>Existing classification*</b>	<p>Met. Corr. 1 (H290)</p> <p>Acute Tox. 4 (H302)</p> <p>Skin Sens 1 (H317)</p> <p>Eye Dam. 1 (H318)</p> <p>Aquatic Acute 1 (H400)</p> <p>Aquatic Chronic 1 (H410)</p> <p>Acute M-factor 100</p> <p>Chronic M-factor 10</p>
<b>Registration deadline</b>	2018

\* For the pure form, as in the REACH registration dossier

### 10. Scope of the Registration Dossier

The uses included in this Registration Dossier are listed on the [EPMF website](#).

## 11. Analytical reference information

Below the results of Raman (Palladium dichloride solution) and XRD (Palladium dichloride solid) analyses of a reference sample used for testing.

### Raman spectroscopy

Apparatus: Raman WITec Alpha 300R

Nd: YAG Laser Compass 315-50 (532 nm)

Sample preparation: For measurement the neat test item (stored in a glass vial) was positioned in a holder of the macro sampling set. After maximizing the signal intensity of the test item's Raman bands at 275 cm<sup>-1</sup> and at 331 cm<sup>-1</sup>, respectively, a Raman spectrum was recorded.

Test parameters: Spectral range: 98.86 cm<sup>-1</sup> - 3649.78 rel. cm<sup>-1</sup>

Resolution: < 6 cm<sup>-1</sup> (not linear)

Excitation wavelength: 532.260 nm

Grating: T1: 600grids/mm BLZ=500 nm

No. of accumulations: 60

Integration time: 1.00002s

Lens: Renishaw Macro Sampling Set, (90° adaptor, lens f = 30mm NA = 0.17)

Measurement at room temperature.

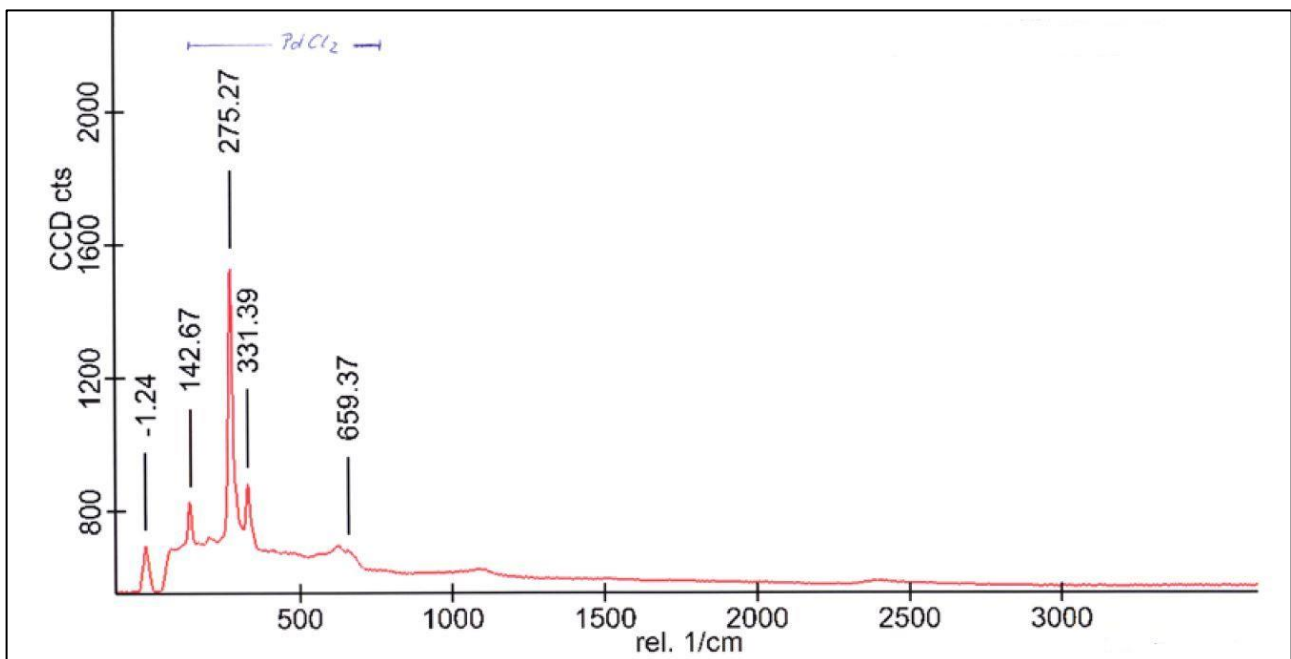


Figure 1. Raman spectrum of Palladium dichloride solution

### X-ray diffraction

Apparatus: X-ray diffractometer Bruker AXS, D8 Advance, Cu tube

Sample preparations: The test item is homogenized in a mortar, put into the sample holder with a glassinlay, smoothed with a glass plate. Due to the small amount of palladium dichloride the sample was glued into the holder with Oppanol.

Test parameters: 2-80° (2 $\theta$ ), 0.02° step size, 1.2 s step time

Primary side: divergence slit: 0.1° with ASS

Secondary side: Lynx-Eye detector with 3° slit

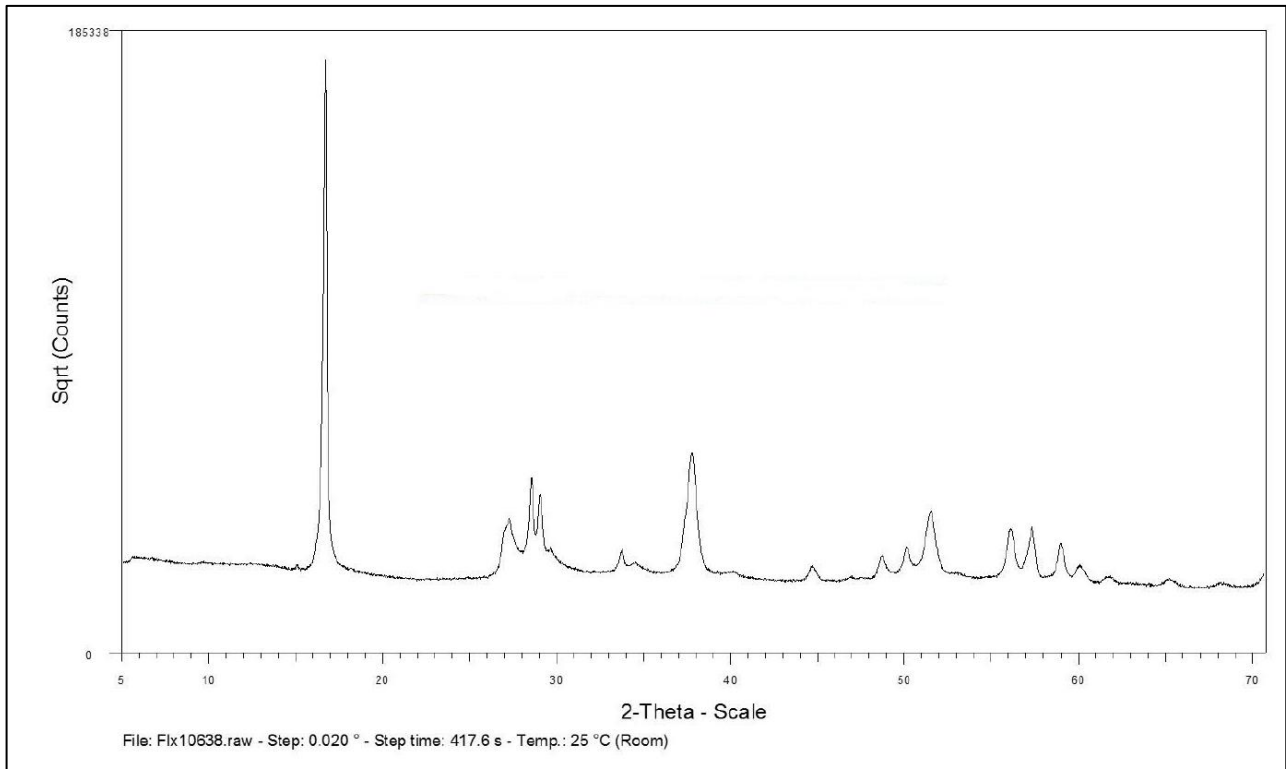


Figure 2. XRD spectrum of Palladium dichloride solid