



ID Card

Ruthenium trichloride (hydrate)

Version 4 July 2023

Notes:

- This ID card is used to support the substance sameness discussions and to describe the substance to the best of the members' knowledge.
- It also aims at grouping communications relevant to the request of available data or information, the approval of the proposed Lead Registrant and the registration strategy.
- It is the responsibility of each individual registrant to identify their substance and to report company-specific identity in their Registration Dossier (section 1 of IUCLID).

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1. Identification of the substance

Table 1. Identification of the substance

	Original (in EC inventory)
Name	Ruthenium trichloride hydrate
EC number	604-667-4
CAS number	14898-67-0
Description	Not available
Composition type	Mono-constituent substance

2. Synonyms and other identifiers of the substance

Table 2. Synonyms and other identifiers of the substance

IUPAC name	Trichlororuthenium hydrate (1:1)
CAS name	
Abbreviations	
Other commercial, brand or international names	Ruthenium(III) chloride hydrate
Other identity codes	

3. Substances (with core identifiers) also falling under this substance (with justification)

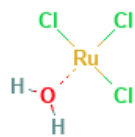
Name	EC number	CAS number	Justification
Ruthenium trichloride trihydrate		13815-94-6	Ruthenium trichloride (hydrate) is a mixture of different hydrates ($\text{RuCl}_3 \cdot x\text{H}_2\text{O}$ with $x=1-3$). The substance is registered under the CAS number of the monohydrate, but the registration also covers the trihydrate.

Note: Under REACH, it is generally recognized that the anhydrous and hydrated forms of a substance are in most cases regarded as the same substance (despite potentially having different CAS numbers). As such, they are mostly covered by the same REACH dossier. However, experts of the EPMF concluded during substance ID discussions that Ruthenium trichloride is an exception to this rule: Ruthenium trichloride (hydrate) is not the same substance as Ruthenium trichloride (anhydrous). The 'hydration water' in Ruthenium trichloride (hydrate) is not bound via hydrogen bonds to the metal ion, as is the case in 'common' hydrates, but is coordinated to the metal to form water soluble chloro-aqua complexes. The anhydrous forms are very low solubility chloro-bridged polymers/oligomers. Because of this different chemical structure of the hydrated vs anhydrous form, the substance properties differ and both forms need registration in different REACH dossiers.

It was confirmed by the EPMF membership that only the hydrated form is currently of commercial relevance and requires REACH registration.

4. Information related to molecular and structural formula of the substance

Table 3. Information related to molecular and structural formula of the substance

Molecular formula	$\text{RuCl}_3 \cdot x\text{H}_2\text{O}$ with x ranging from 1-3
Structural formula	
Smiles notation	<chem>O.Cl[Ru](Cl)Cl</chem>
Optical activity	Not applicable
Typical ratio of (stereo) isomers	Not applicable
Molecular Weight / Molecular Weight range	225,44 g/mol (monohydrate) – 261.48 g/mol (trihydrate)

5. Typical composition of the substance

Table 4. Typical composition

	Name	Symbol / Formula	Min & Max concentrations (%) [§]	Typical concentration (%) ^{§§}
Main constituent(s)*	Ruthenium trichloride hydrate	RuCl ₃ .xH ₂ O with x=1-3	89-100 [§]	96
Impurities[#]	HCl		0 – 10	3.3
	Several minor (especially metallic) impurities which do not affect the classification of the substance because of their non-hazardous nature or because they do not exceed the classification cut-off limits in the substance	e.g. Ag, Au, Cu, Ir, Pd, Pt, Rh	0 – 1	< 0.7

* ≥ 80 % (w/w) for mono-constituent substances; ≥ 10 % (w/w) and < 80 % (w/w) for multi-constituent substances.

[#] An impurity is an unintended constituent present in a substance, as produced. It may originate from the starting materials or be the result of secondary or incomplete reactions during the production process. While impurities are present in the final substance, they were not intentionally added.

[§] Concentration ranges define the substance sameness criteria agreed by all EPMF Members in preparation of the communication with other SIEF members.

^{§§} Typical concentration refers to the representative sample used for testing.

[§] Corresponds to 34,4 - 45,1 % Ru.

The composition given above is typical and should therefore represent the majority of Ruthenium trichloride (hydrate) as manufactured and/or imported in the EEA market. Ruthenium trichloride (hydrate) containing less Ruthenium trichloride (hydrate) may still be considered to be the same for the purpose of registration under REACH and may be referred to as impure Ruthenium trichloride (hydrate) to distinguish it from the typically pure Ruthenium trichloride (hydrate).

6. Information on appearance, physical state and properties of the substance

Table 5. Appearance / physical state / properties of the solid substance

Physical state	Solid
Physical form*	Crystalline
Appearance	Black powder
Particle size**	Fine to coarse powder
Does the solid hydrolyse?#	Yes / No
Is the solid hygroscopic?§	Yes / No

* Crystalline form: solid material whose constituent atoms, molecules, or ions are arranged in an ordered pattern extending in all three spatial dimensions. Amorphous form: solid material whose constituent atoms, molecules, or ions are randomly arranged.

** Nanoform: particles in the size range 1 - 100 nm (for full definition of a nanomaterial, see <http://ec.europa.eu/environment/chemicals/nanotech/index.htm#definition>). Fine powder: particles in the size range 100 – 2.500 nm. Coarse powder: particles in the size range 2.500 nm – 1 mm. Massive object: particles in the size range > 1 mm.

Hydrolysis: decomposition (cleavage of chemical bonds) by the addition of water.

§ Hygroscopic substance: readily attracts moisture from its surroundings in open air, through either absorption or adsorption. Cf. also water/moisture content in Table 4.

7. Analytical data

Annex VI of REACH requires the registrant to describe the analytical methods and/or to provide the bibliographical references for the methods used for identification of the substance and, where appropriate, for the identification of impurities and additives. This information should be sufficient to allow the methods to be reproduced.

Table 6. Analytical methods for identification of the substance

Parameter / Method	Recommended for substance identification and sameness check	Applicable	Not applicable or not recommended
Elemental analysis			
ICP (ICP-MS or ICP-OES)	X		
Atomic absorption spectroscopy (AAS)			
Glow discharge mass spectrometry (GDMS)			
Molecular analysis			
Infrared (IR) spectroscopy	X		
Raman spectroscopy			
Mineralogical analysis			
X-Ray Fluorescence (XRF)		X	
X-Ray Diffraction (XRD)	X		
Morphology and particle sizing			
Electron microscopy (SEM, TEM, REM)* #			
Laser diffraction* #	X		
Particle size by other means (e.g. sieve analysis)#			
Surface area by N-BET* #	X		
Other			

* Analytical techniques particularly (but not exclusively) relevant for nanomaterials.

The choice of the technique for particle size depends on the size of the material as manufactured/imported/placed on the market/used.

8. Lead Registrant

Heraeus Deutschland GmbH & Co. KG (Germany) is the Lead Registrant for Ruthenium trichloride (hydrate). The EPMF will provide support to the Lead Registrant as laid down in the EPMF Agreement.

9. Scope of the Registration Dossier

The uses included in this Registration Dossier are listed on the [EPMF website](#).

10. Analytical reference information

Below the results of IR analysis of a reference sample used for testing.

Spectrometer: Infrared spectrometer Tensor 27

Spectral Range: 4500 – 200 cm⁻¹

Resolution: 2 cm⁻¹

Scans: 32

Temperature: ambient

Sample preparation: Nujol mull, undefined layer CsJ

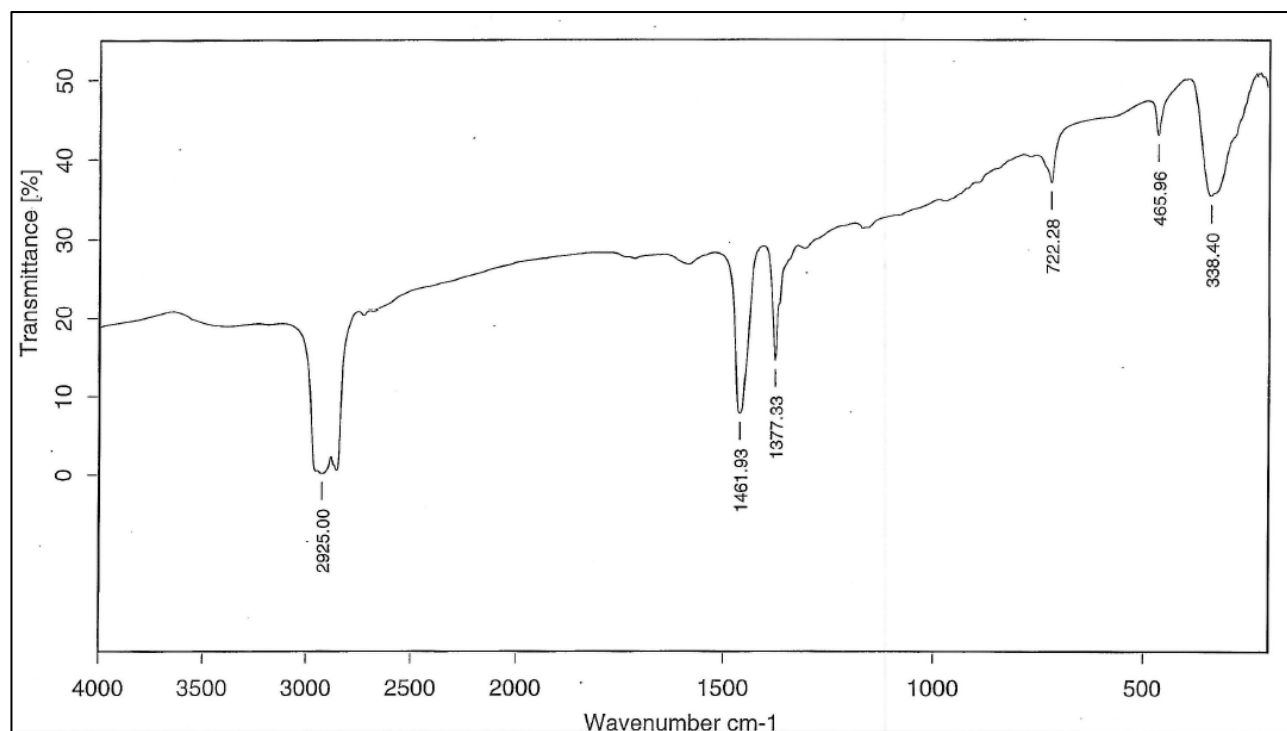


Figure 1. IR spectrum of Ruthenium trichloride (hydrate)